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point compared to oleic acid, $\text{CH}_3(\text{CH}_2)_7\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$. Outline your reason. [2]..... B3. (a) Some synthetic hormones are similar in structure to progesterone and estrogen and may be used to

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prevent pregnancy. Outline the mode of action of these hormones as oral contraceptives. [3].....

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HL Paper 3 Infrared (IR) spectroscopy is widely used as a technique in analytical chemistry. The IR spectrum, mass spectrum and spectrum of an unknown compound, Y, of molecular formula are as follows. (i) Identify the bonds responsible for the peaks A, B and C in the IR spectrum of Y. A: B: C:

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3. a ionic 1 3. b i red 1 3. b ii emission spectra of both « ^6Li and natural Li» give same colour/produce same «range of» wavelengths OR they have same electron transitions/same nuclear charge Accept “the spectra are almost identical”. 1 3. b iii ICP -MS Accept “MS/mass spectrometry”. 1 3. c r m 0.694 n 0.100 mol M 6.94 «

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(vi) $\text{HCOOC}(\text{CH}_3)_3$ is an isomer of X. For the ^1H NMR spectrum of this isomer, deduce the total number of peaks (excluding the TMS peak at 0 ppm) and the ratio of peak areas. For each peak, deduce whether it is a singlet, doublet, triplet, quartet or shows a more complex splitting pattern. [3] Number of peaks (excluding the TMS peak at 0 ppm):

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IB Chemistry SL and HL Core. Both IB Chemistry SL and HL cover the first 11 topics (a total of 95 hours), and HL additionally covers topics 12 to 21 (an additional 60 hours). For both levels, you'll also study one of the four options A-D (15 hours for SL, 25 hours for HL). Topic #1: Stoichiometric Relationships—13.5 Hours for SL and HL

The Complete IB Chemistry Syllabus: SL and HL

volume made up to the graduated mark with distilled water. A 10.0 cm³ sample of this new solution was pipetted into a new conical flask and titrated with 0.100 mol dm⁻³ sodium hydroxide solution using phenolphthalein as the indicator. 12.60 cm³ of the 0.100 mol dm⁻³ sodium hydroxide solution was

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required to reach the end point.

SL & HL Paper 3 Section A Experimental work Question 7

The 1:2 ratio of marks (paper 3-to-paper 2) for SL versus the 2:3 ratio for HL is important to bear in mind when deciding how much time to give over to each exam in preparation. On a similar note, the emphasis on the Option is different for SL and HL. The Option forms a much smaller fraction of the overall mark if you are studying it at SL.

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Advances in Clinical Chemistry

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have taken account of many developments in the field, culminating in the major extension and revision represented by the 1988 edition under the simplified title *Quantities, Units and Symbols in Physical Chemistry*. This 2007, Third Edition, is a further revision of the material which reflects the experience of the contributors with the previous editions. The book has been systematically brought up to date and new sections have been added. It strives to improve the exchange of scientific information among the readers in different disciplines and across different nations. In a rapidly expanding volume of scientific literature where each discipline has a tendency to retreat into its own jargon this book attempts to provide a readable compilation of widely used terms and symbols from many sources together with brief understandable definitions. This is the definitive guide for scientists and organizations working across a multitude of disciplines requiring internationally approved nomenclature.

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